

LOWELL OBSERVATORY  
FLAGSTAFF, ARIZONA

April 25, 1959

Dr. Joshua Lederberg  
Department of Genetics  
School of Medicine  
Stanford University  
Stanford, California

Dear Dr. Lederberg:

Here is the analysis of the feasibility of detecting organic matter on Mars from a probe.

First the detection from a probe of NH bending and C=O stretching bands at 5 - 6  $\mu$  is considered. Most of the energy from Mars at these wavelengths will be thermal emission. I find that for an emissivity of 0.5 the energy received per  $\text{cm}^2$  from the whole of Mars at 100,000 miles within a 0.1  $\mu$  band is  $5 \times 10^{-9}$  watts. Since lead selenide detectors at liquid nitrogen temperatures can detect  $10^{-9}$  watts or less, from a detection standpoint the test seems feasible. Telescope apertures of the order of 100  $\text{cm}^2$  would seem to yield adequate signal-to-noise ratios. The bands will appear in emission since, as mentioned, the majority of the energy is thermal emission. Some amount of surface discrimination may be possible, particularly if larger telescope apertures can be used. Observation of these bands from the ground is prohibited poor atmospheric transmission and the small amount of energy. From a balloon this latter difficulty will be considerably enhanced by restricted apertures available.

Another possibility I have considered is detection of N-H and O-H stretching bands at 3  $\mu$ . This region is again messed up at ground level by a water band. In a balloon this difficulty is eliminated. One expects about  $2 \times 10^{-11}$  watts/0.1  $\mu$  from the whole of Mars in a 12-inch aperture. This would be detectable by using a cooled lead sulfide cell. To obtain resolution on the surface a considerably larger aperture or a probe is required.

I have learned, partly to my chagrin, that carbonates have prominent bands right at 3.4 microns. However, the presence of carbonates on Mars is eliminated by the fact that their bands are not like the observed in that there is not appreciable absorption at 3.67  $\mu$ . It is still further eliminated in that carbonates have a very strong absorption at 3.9  $\mu$  which is not observed on Mars. However, I feel that, from the observations alone, one can not be sure that the observed absorptions are produced by organic molecules.

I have enclosed the travel vouchers and my tickets.

Sincerely,

Bill

William M. Sinton

Sinton, W.M.